

In the Claims

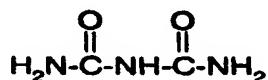
Please amend the claims as detailed herein below:

1. (Original) A method for preparing hydrazodicarbonamide which comprises the steps of:

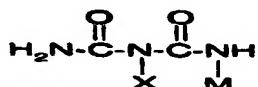
obtaining biuret of Formula 1 and ammonia by pyrolysis of urea;

obtaining metal monohalobiuret salt of Formula 2 or 3 by reacting the obtained biuret with metal hypohalogen compound or with halogen and base; and reacting the obtained metal monohalobiuret salt with ammonia including ammonia which is produced from the urea pyrolysis process, and the mole ratio of the metal monohalobiuret salt and the total ammonia is 1:1 -1 :1000,

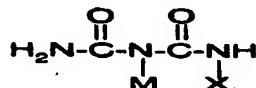
[Formula 1]



[Formula 2]



[Formula 3]



In Formula 2 and 3, M represents metal and X represents halogen.

2. (Original) The method for preparing hydrazodicarbonamide according to claim 1, wherein the temperature of pyrolysis of urea is 100 to 300°C.
3. (Original) The method for preparing hydrazodicarbonamide according to claim 1,

wherein the urea pyrolysis process is carried out with removing ammonia from the reaction system thereof.

4. (Currently Amended) The method for preparing hydrazodicarbonamide according to claim 1, wherein the urea pyrolysis process is carried out with a process of injection of inert gas ~~and/or with a process of reducing pressure of the reaction system thereof.~~
5. (Currently Amended) The method for preparing hydrazodicarbonamide according to claim 1, wherein the urea pyrolysis process is carried out in the presence of at least one catalyst selected from the group consisting of inorganic acid catalyst, ~~and acid type catalyst, and substances including phosphorous.~~
6. (Original) The method for preparing hydrazodicarbonamide according to claim 1, wherein the mole ratio of the biuret and metal hypohalogen compound is 1:0.1~1:2.
7. (Original) The method for preparing hydrazodicarbonamide according to claim 1, wherein the metal monohalobiuret salt is obtained by mixing metal hydroxide with the biuret of Formula 1, and then reacting the obtained product with halogen, or obtained by reacting the biuret of Formula 1 with gaseous halogen, and then mixing the obtained product with a base.
8. (Original) The method for preparing hydrazodicarbonamide according to claim 1, wherein the process of obtaining the metal monohalobiuret salt is carried out at a temperature of below 60°C.
9. (Original) The method for preparing hydrazodicarbonamide according to claim 1, wherein the ammonia is liquid ammonia, gaseous ammonia or ammonium hydrate.

10. (Original) The method for preparing hydrazodicarbonamide according to claim 1, wherein the reaction of the metal monohalobiuret salt with the ammonia is carried out at a temperature of 0 to 150°C.

11. (Currently Amended) The method for preparing hydrazodicarbonamide according to claim 1, wherein the solvent of the ~~reaction~~ biuret is a mixture of water and a second solvent, and the second solvent is a polar solvent selected from the group consisting of water, methanol, ethanol, propanol, isopropanol and mixtures thereof, or a aprotic solvent selected from the group consisting of dimethylformamide, dimethylsulfoxide, dimethylacetamide and mixtures thereof.

12. – 14. (Canceled)